

Poly[zinc(II)- μ_2 -4,4'-bipyridine-di- μ_2 -formato]: a chiral three-dimensional coordination polymer

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Key indicators

Single-crystal X-ray study
T = 295 K
Mean $\sigma(\text{C}-\text{C}) = 0.002 \text{ \AA}$
R factor = 0.019
wR factor = 0.059
Data-to-parameter ratio = 13.7

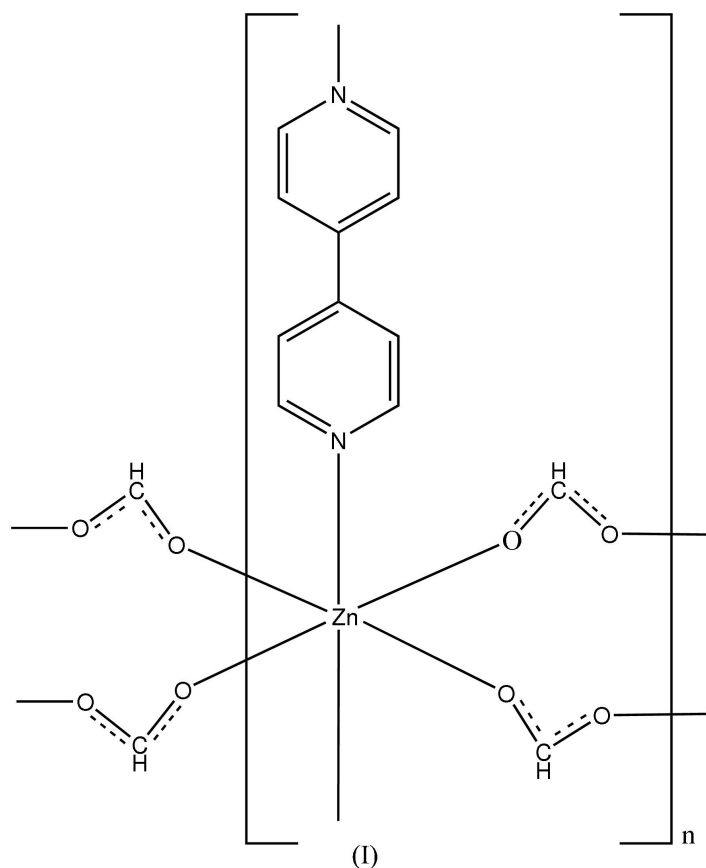
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The title compound, $[\text{Zn}(\text{CHO}_2)_2(\text{C}_{10}\text{H}_8\text{N}_2)]_n$, is a chiral three-dimensional coordination polymer. The Zn atom, the two bipyridine N atoms, and the bridging bipyridine C atoms occupy special positions with site symmetry 2. In the extended structure, both the ligands act as linkers in μ_2 -mode. The compound is isostructural with its copper analogue.

Received 3 July 2006
Accepted 4 July 2006

Comment

The reaction of copper formate and sodium formate with 4,4'-bipyridine (4,4'-bipy) in acetone solution yielded a chiral three-dimensional coordination polymer, $[\text{Cu}(\text{HCOO})_2(\text{C}_{10}\text{H}_8\text{N}_2)]_n$ (Manson *et al.*, 2003). Unexpectedly, the reaction of zinc acetate, sodium hydrogen 3-sulfobenzoate and 4,4'-bipyridine in the mixed-solvent system, water and *N,N'*-dimethylformamide under reflux conditions led to the title compound, (I), which is isostructural with its copper analogue (Manson *et al.* 2003).



The Zn^{II} atom in (I) adopts an octahedral geometry coordinated by two *trans* N-atom donors from two 4,4'-bipyridine

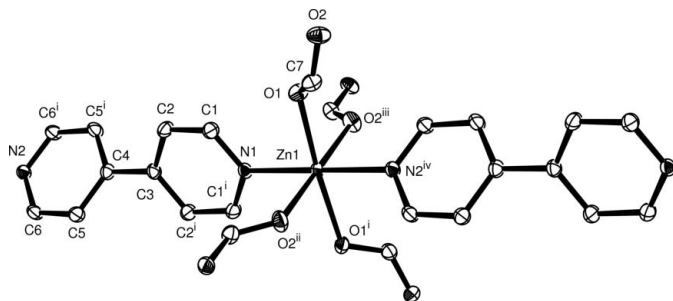


Figure 1

View of a fragment of the polymeric structure of (I), with displacement ellipsoids drawn at the 50% probability level (H atoms omitted for clarity). [Symmetry codes: (i) $y, x, -z$; (ii) $-\frac{1}{2} + x, \frac{1}{2} - y, -\frac{1}{4} - z$; (iii) $\frac{1}{2} - y, x - \frac{1}{2}, \frac{1}{4} + z$; (iv) $1 + x, 1 + y, z$.]

molecules and four O atoms from four formate ligands (Fig. 1 and Table 1). Atoms Zn1, N1, C3, C4 and N2 occupy special positions with site symmetry 2. The $[\text{Zn}(4,4'\text{-bipy})]$ units give rise to one-dimensional chains propagating alternately in $[110]$ and $[\bar{1}10]$. Within a chain, the two pyridyl rings of each 4,4'-bipyridine molecule are twisted with a dihedral angle of $47.67(6)^\circ$. The $\text{Zn} \cdots \text{Zn}$ separation *via* the 4,4'-bipyridine bridge is $11.2694(4) \text{ \AA}$. The formate ligands act as bridges in μ_2 -mode and extend the $[\text{Zn}(4,4'\text{-bipy})]$ chains into a three-dimensional network. The formate C—O distances suggest delocalization of the negative charge of this ion.

Experimental

A mixture of $\text{Zn}(\text{CH}_3\text{COO})_2 \cdot 4\text{H}_2\text{O}$ (0.112 g, 0.44 mmol), sodium hydrogen 3-sulfobenzoate (0.112 g, 0.5 mmol), 4,4'-bipyridine (0.078 g, 0.5 mmol), water (30 ml) and *N,N*-dimethylformamide (5 ml) was refluxed for 7 h. The resulting solution was set aside and allowed to evaporate. After two months, colorless crystals of (I) were obtained.

Crystal data

$[\text{Zn}(\text{CHO}_2)_2(\text{C}_{10}\text{H}_8\text{N}_2)]$
 $M_r = 311.59$
 Tetragonal, $P4_12_12$
 $a = 7.9687(3) \text{ \AA}$
 $c = 17.7213(11) \text{ \AA}$
 $V = 1125.31(9) \text{ \AA}^3$
 $Z = 4$

$D_x = 1.839 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 $\mu = 2.19 \text{ mm}^{-1}$
 $T = 295(2) \text{ K}$
 Block, colorless
 $0.39 \times 0.30 \times 0.22 \text{ mm}$

Data collection

Bruker APEX CCD diffractometer
 φ and ω scans
 Absorption correction: multi-scan
 (SADABS; Bruker, 2002)
 $T_{\min} = 0.452, T_{\max} = 0.613$

6797 measured reflections
 1232 independent reflections
 1206 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.024$
 $\theta_{\max} = 27.0^\circ$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.019$
 $wR(F^2) = 0.059$
 $S = 0.87$
 1232 reflections
 90 parameters
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0566P)^2 + 0.0576P]$
 where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.28 \text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.34 \text{ e \AA}^{-3}$
 Extinction correction: SHELXL97
 Extinction coefficient: 0.0145 (15)
 Absolute structure: Flack (1983),
 451 Friedel pairs
 Flack parameter: 0.008 (13)

Table 1

Selected bond lengths (\AA).

| | | | |
|---------------------|-------------|----------------------|-------------|
| Zn1—N1 | 2.1065 (17) | Zn1—O2 ⁱⁱ | 2.1831 (11) |
| Zn1—N2 ⁱ | 2.1204 (16) | C7—O1 | 1.2535 (17) |
| Zn1—O1 | 2.1729 (11) | C7—O2 | 1.2373 (18) |

Symmetry codes: (i) $x + 1, y + 1, z$; (ii) $x - \frac{1}{2}, -y + \frac{1}{2}, -z - \frac{1}{4}$.

All H atoms were positioned geometrically ($\text{C—H} = 0.93 \text{ \AA}$) and refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: SMART (Bruker, 2002); cell refinement: SAINT (Bruker, 2002); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

The authors thank the National Natural Science Foundation of China (grant No. 50073019), and the Analytical and Measurement Fund of Zhejiang Province.

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